

Counting Total Number of Microstates of the Combined System in the Microcanonical Ensemble

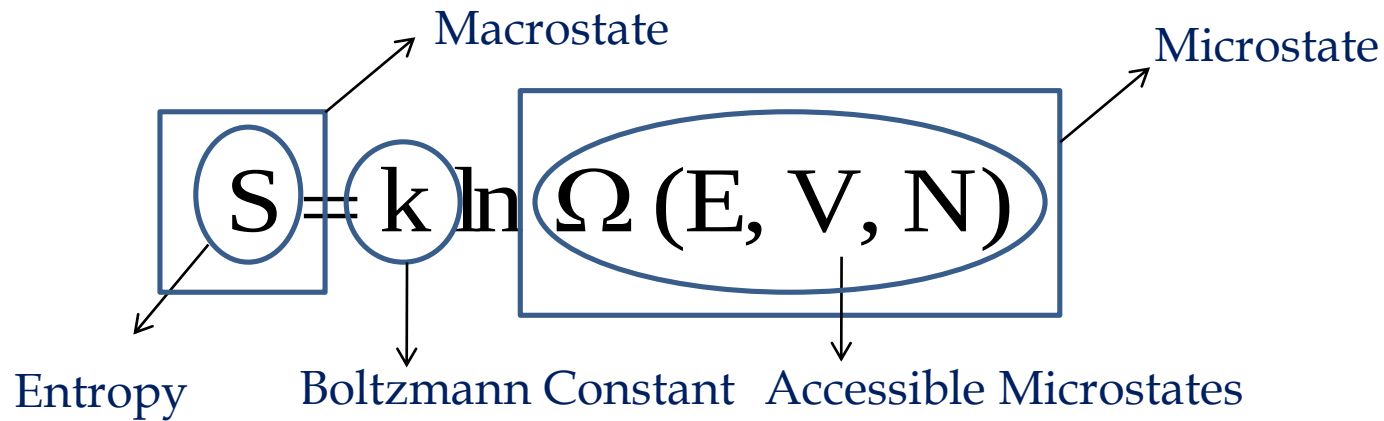
The Subsystem 'A' has $\Omega_A = 6$ accessible *Microstates*.

The Subsystem 'B' has $\Omega_B = 2$ accessible *Microstates*.

Total No of *Microstates* Ω_{tot} of the composite system.

$$\Omega_{\text{tot}} = \Omega_A \times \Omega_B = 12$$

The Partition prevents the transfer of energy from one subsystem to another and in this case keeps $E_A = 5$ and $E_B = 1$.



Thermodynamics

$$\frac{1}{T} = \left(\frac{\partial S}{\partial E} \right)_{V, N}$$

$$\frac{P}{T} = \left(\frac{\partial S}{\partial V} \right)_{E, N}$$

$$\frac{\mu}{T} = - \left(\frac{\partial S}{\partial N} \right)_{E, V}$$

Canonical Ensemble: Counting the Number of **Microstates**

The Microstates are given in the table for all the possible values of E_A and E_B

E_A	E_B	Microstates of E_A	Microstates of E_B	$\Omega_A(E_A)$	$\Omega_A(E_B)$	$\Omega_A \Omega_B$	$P_A(E_A)$
6	0	(6,0) (5,1) (4,2) (3,3) (2,4) (1,5) (0,6)	(0,0)	7	1	7	7/84
5	1	(5,0) (4,1) (3,2) (2,3) (1,4) (0,5)	(1,0) (0,1)	6	2	12	12/84
4	2	(4,0) (3,1) (2,2) (1,3) (0,4)	(2,0) (1,1) (0,2)	5	3	15	15/84
3	3	(3,0) (0,3) (2,1) (1,2)	(3,0) (0,3) (2,1) (1,2)	4	4	16	16/84
2	4	(2,0) (1,1) (0,2)	(4,0) (3,1) (2,2) (1,3) (0,4)	3	5	15	15/84
1	5	(1,0) (0,1)	(5,0) (4,1) (3,2) (2,3) (1,4) (0,5)	2	6	12	12/84
0	6	(0,0)	(6,0) (5,1) (4,2) (3,3) (2,4) (1,5) (0,6)	1	7	7	7/84
2/17/2024						$\Omega_{\text{tot}} = 84$	

The Microstates are given in the table for all the possible values of E_A and E_B

E_A	E_B	Microstates of E_A	Microstates of E_B	$\Omega_A(E_A)$	$\Omega_A(E_B)$	$\Omega_A \Omega_B$	$P_B(E_B)$
6	0	(6,0) (5,1) (4,2) (3,3) (2,4) (1,5) (0,6)	(0,0)	7	1	7	7/84
5	1	(5,0) (4,1) (3,2) (2,3) (1,4) (0,5)	(1,0) (0,1)	6	2	12	12/84
4	2	(4,0) (3,1) (2,2) (1,3) (0,4)	(2,0) (1,1) (0,2)	5	3	15	15/84
3	3	(3,0) (0,3) (2,1) (1,2)	(3,0) (0,3) (2,1) (1,2)	4	4	16	16/84
2	4	(2,0) (1,1) (0,2)	(4,0) (3,1) (2,2) (1,3) (0,4)	3	5	15	15/84
1	5	(1,0) (0,1)	(5,0) (4,1) (3,2) (2,3) (1,4) (0,5)	2	6	12	12/84
0	6	(0,0)	(6,0) (5,1) (4,2) (3,3) (2,4) (1,5) (0,6)	1	7	7	7/84

- The Total Number of Microstates $\Omega_{\text{tot}}(E_A + E_B)$ accessible to the isolated composite system whose subsystems have energy E_A and E_B is

$$\bullet \quad \Omega_{\text{tot}}(E_A + E_B) = \Omega_A(E_A) \Omega_B(E_B)$$

- For example if $E_A = 4$ and $E_B = 2$, then 'A' can be any one of the 5 microstates and 'B' can be in an one of 3 microstates. There are two sets of microstates that can be combined to give 15 microstates.
- The Total Number of Microstates Ω_{tot} accessible to the composite system can be found by summing $\Omega_A(E_A) \Omega_B(E_B)$ for the possible ways of E_A and E_B consistent with the condition $E_A + E_B = 6$.

$$\begin{aligned} \Omega &= \sum_{E_A} \Omega_A(E_A) \Omega_B(E_B) \\ &= \sum \Omega_A(E_A) \Omega_B(E_{\text{tot}} - E_A) \end{aligned}$$

From Table,

- $\Omega_{\text{tot}} = (7 \times 1) + (6 \times 2) + (5 \times 3) + (4 \times 4) + (3 \times 5) + (2 \times 6) + (1 \times 7)$
 $= 84$

The *Composite System* is isolated. So all accessible *Microstates* are equally probable.

That is the composite is equally likely to be in any of its 84 accessible *Microstates*.

- Let $P_A(E_A)$ = Probability that subsystem 'A' has energy E_A .

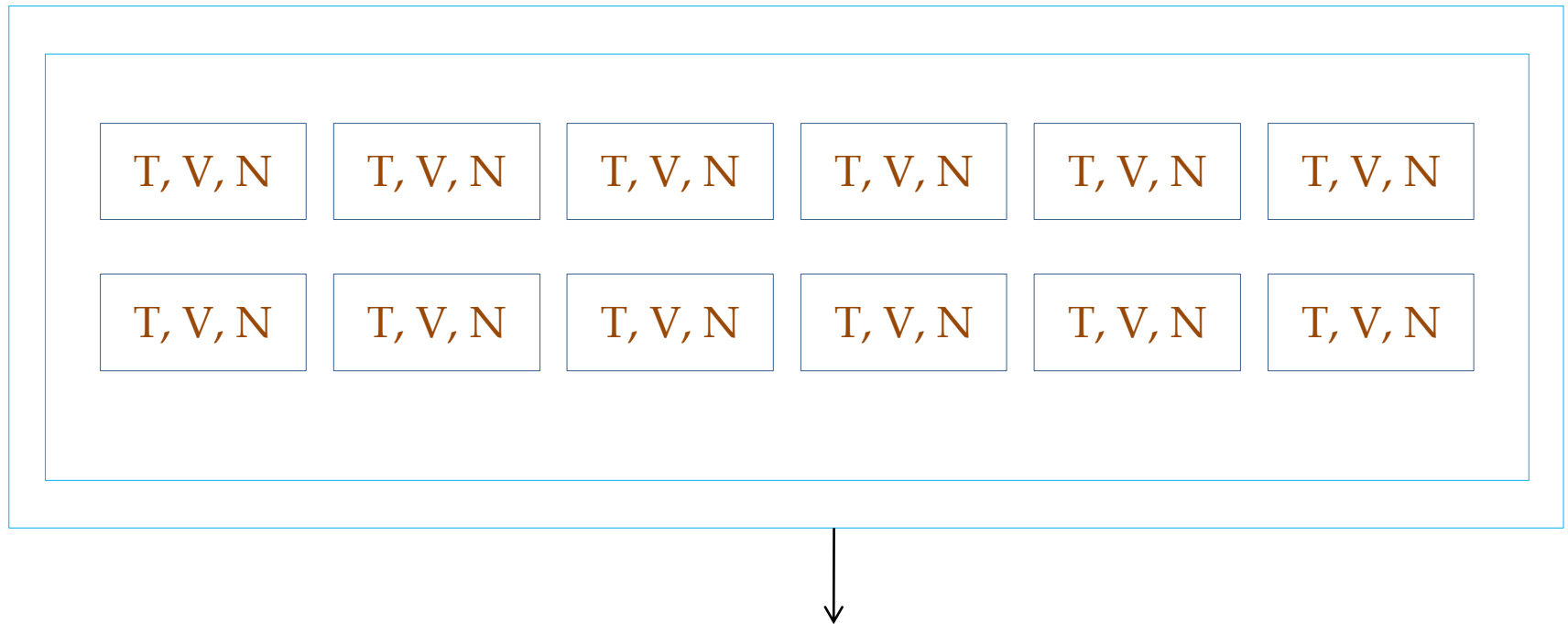
$$P_A(E_A) = \frac{\sum \Omega_A(E_A) \Omega_B(E_{\text{tot}} - E_A)}{\Omega_{\text{tot}}}$$

- The mean energy of subsystem 'A' is found by calculating the *Ensemble* average over the 84 *Microstates* of the *Composite System*.

$$\begin{aligned} \langle E_A \rangle &= \bar{E}_A = \sum E_{A,i} P(E_{A,i}) \\ &= \left(6 \times \frac{7}{84} \right) + \left(5 \times \frac{12}{84} \right) + \left(4 \times \frac{15}{84} \right) + \left(3 \times \frac{16}{84} \right) + \left(2 \times \frac{15}{84} \right) \\ &\quad + \left(1 \times \frac{12}{84} \right) + \left(0 \times \frac{7}{84} \right) \\ &= 3 \end{aligned}$$

Hence, it is reasonable to assume that when a closed isolated system had time to reach internal *Thermodynamic Equilibrium*, we are likely to find the closed system in any one of the microstates.

At *Thermal Equilibrium*, the probability of finding the *Composite System* in any one of the **84** accessible *Microstates* of the *Composite System* is **1/84**.



Canonical Ensemble

In this example, the **Ensemble** consists of 84 systems each of which is in one of the 84 accessible Microstates.

Probability of finding the particle in a particular microstate

$$P_i = \frac{e^{-\beta\epsilon_i}}{\sum e^{-\beta\epsilon_i}} = \frac{e^{-\beta\epsilon_i}}{Z}$$

$$Z = \sum e^{-\beta\epsilon_i}$$

→ Partition function

- ❖ The partition function '**Z**' contains all the information about the energies of the states of the system.
- ❖ All thermodynamical quantities can be obtained from it.
- ❖ From '**Z**', we can get thermodynamics functions energy, entropy, Helmholtz function, or heat capacity to simply drop out.

Thermodynamic quantities derived from the partition function Z

Function of states	Statistical mechanical expression
U F	$-\frac{d \ln Z}{d\beta}$ $-k_B T \log Z$
$S = -\left(\frac{\partial F}{\partial T}\right)_V = \frac{U - F}{T}$ $p = -\left(\frac{\partial F}{\partial V}\right)_T$ $H = U + pV$ $G = F + pV = H - TS$ $C_V = \left(\frac{\partial U}{\partial T}\right)_V$	$k_B \ln Z + k_B T \left(\frac{\partial \ln Z}{\partial T}\right)_V$ $k_B T \left(\frac{\partial \ln Z}{\partial V}\right)_T$ $k_B T \left[T \left(\frac{\partial \ln Z}{\partial V}\right)_V + V \left(\frac{\partial \ln Z}{\partial V}\right)_T \right]$ $k_B T \left[-\ln Z + V \left(\frac{\partial \ln Z}{\partial V}\right)_T \right]$ $k_B T \left[2 \left(\frac{\partial \ln Z}{\partial T}\right)_V + T \left(\frac{\partial^2 \ln Z}{\partial T^2}\right)_V \right]$