## Amino acids

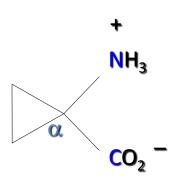
### Prof. Narkunaraja Shanmugam

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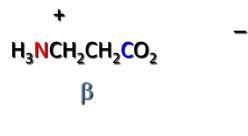
## **Fundamentals**

- While their name implies that amino acids are compounds that contain an  $-NH_2$  group and a  $-CO_2H$  group, these groups are actually present as  $-NH_3^+$  and  $-CO_2^-$  respectively.
- They are classified as  $\alpha$ ,  $\beta$ ,  $\gamma$ , etc. amino acids according the carbon that bears the nitrogen.

# **Amino Acids**



an  $\alpha$ -amino acid that is an intermediate in the biosynthesis of ethylene



a β-amino acid that is one of the structural units present in coenzyme A

 $_{_{3}\text{NCH}_{2}\text{CH}_{2}\text{CO}_{2}}^{+}$ 

a  $\gamma$ -amino acid involved in the transmission of nerve impulses

# The 20 Key Amino Acids

- More than 700 amino acids occur naturally, but 20 of them are especially important.
- These 20 amino acids are the building blocks of proteins. All are  $\alpha$ -amino acids.
- They differ in respect to the group attached to the  $\alpha$  carbon.
- These 20 are listed in Table

ABLE 27.1 α-Amino Acids Found in Proteins			
Name	Abbreviation	Structural formula*	
Amino acids with nonpolar side chains			
Glycine	Gly (G)	<sup>+</sup> NH₃ 	
Alanine	Ala (A)	NH₃   H₃C—CHCO₂ <sup>−</sup>	
Valine <sup>†</sup>	Val (V)	$\stackrel{NH_3}{\mid}$ $(CH_3)_2CH - CHCO_2^-$	
Leucine <sup>†</sup>	Leu (L)	$(CH_3)_2CHCH_2$ — $CHCO_2$	
Isoleucine <sup>†</sup>	lle (I)	CH <sub>3</sub> NH <sub>3</sub>       CH <sub>3</sub> CH <sub>2</sub> CH—CHCO <sub>2</sub>	
Methionine <sup>†</sup>	Met (M)	CH <sub>3</sub> SCH <sub>2</sub> CH <sub>2</sub> —CHCO <sub>2</sub> —	
Proline	Pro (P)	$H_2C$ $\uparrow$ $H_2$ $H_2C$ $\downarrow$	
Phenylalanine <sup>†</sup>	Phe (F)	~	
Tryptophan <sup>†</sup>	Trp (W)	CH <sub>2</sub> —CHCO <sub>2</sub> -	
Amino acids with polar but nonionized side chains			

Asn (N) Asparagine H<sub>2</sub>NCCH<sub>2</sub>—CHCO<sub>2</sub>

(Continued)

Table 27.1  $\alpha$ -Amino Acids Found in Proteins (Continued)

Name	Abbreviation	Structural formula*		
Amino acids with polar but nonionized side chains				
		O NH <sub>3</sub>		
Glutamine	Gln (Q)	O NH <sub>3</sub>      H <sub>2</sub> NCCH <sub>2</sub> CH <sub>2</sub> —CHCO <sub>2</sub> -		
		NH <sub>3</sub>		
Serine	Ser (S)	HOCH <sub>2</sub> —CHCO <sub>2</sub> -		
		OH ŅH₃		
Threonine <sup>†</sup>	Thr (T)	CH₃CH—CHCO₂¯		
		, NH <sub>3</sub>		
Tyrosine	Tyr (Y)	HO—CH <sub>2</sub> —CHCO <sub>2</sub>		
		<sup>+</sup> NH₃		
Cysteine	Cys (C)	HSCH <sub>2</sub> —CHCO <sub>2</sub> -		
Amino acids with acidic	side chains			
		O		
Aspartic acid	Asp (D)	OCCH <sub>2</sub> —CHCO <sub>2</sub>		
		O NH <sub>3</sub>		
Glutamic acid	Glu (E)	OCCH <sub>2</sub> CH <sub>2</sub> —CHCO <sub>2</sub>		
Amino acids with basic s	ide chains			
		NH <sub>3</sub>		
Lysine <sup>†</sup>	Lys (K)	H <sub>3</sub> NCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> —CHCO <sub>2</sub> —		
		$\stackrel{+}{\text{NH}}_2$ $\stackrel{+}{\text{NH}}_3$		
Arginine <sup>†</sup>	Arg (R)	H <sub>2</sub> NCNHCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> —CHCO <sub>2</sub>		
		, NH³		
Histidine <sup>†</sup>	His (H)	N CH <sub>2</sub> -CHCO <sub>2</sub> -		
		N H		

<sup>\*</sup>All amino acids are shown in the form present in greatest concentration at pH 7.  $^\dagger$ An essential amino acid, which must be present in the diet of animals to ensure normal growth.

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#### $\alpha$ -Amino Acids Found in Proteins *(Continued)*

Name	Abbreviation	Structural formula*		
Amino acids with polar but nonionized side chains				
		Ö ŅH³		
Glutamine	Gln (Q)	$\begin{array}{ccc} O & & \stackrel{\leftarrow}{NH_3} \\ \parallel & & \downarrow \\ H_2NCCH_2CH_2 & & CHCO_2^- \end{array}$		
		, NH3		
Serine	Ser (S)	HOCH <sub>2</sub> —CHCO <sub>2</sub> -		
	5-1 (0)			
_, ,		OH NH <sub>3</sub>     CH <sub>3</sub> CH—CHCO <sub>2</sub>		
Threonine <sup>†</sup>	Thr (T)	CH₃CH—CHCO₂		
		NH <sub>3</sub>		
Tyrosine	Tyr (Y)	$HO \longrightarrow CH_2 - CHCO_2^-$		
		+		
		NH <sub>3</sub>		
Cysteine	Cys (C)	HSCH <sub>2</sub> —CHCO <sub>2</sub>		
Amino acids with acidic side chains				
		O NH <sub>3</sub>       COCCH <sub>2</sub> —CHCO <sub>2</sub> —		
Aspartic acid	Asp (D)	¯OCCH <sub>2</sub> —CHCO <sub>2</sub> ¯		
		O NH <sub>3</sub>		
Glutamic acid	Glu (E)	O NH <sub>3</sub>         -OCCH <sub>2</sub> CH <sub>2</sub> —CHCO <sub>2</sub> -		
Amino acids with basic side chains				
		NH <sub>3</sub>		
Lysine <sup>†</sup>	Lys (K)	H <sub>3</sub> NCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> —CHCO <sub>2</sub>		
		, NH <sub>2</sub> , NH <sub>3</sub>		
Arginine <sup>†</sup>	Arg (R)	H <sub>2</sub> NCNHCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> —CHCO <sub>2</sub> —		
	J . ,	4		
re-e-et	115-710	NH <sub>3</sub> 		
Histidine <sup>†</sup>	His (H)	CH <sub>2</sub> —CHCO <sub>2</sub>		
		Н		

#### **Amino Acids**

Amino Acids are the building units of proteins. Proteins are polymers of amino acids linked together by what is called "Peptide bond".

#### **Structure of amino acids:**

Each amino acid has 4 different groups attached to  $\alpha$ - carbon (which is C-atom next to COOH). These 4 groups are : amino group, COOH gp,

Hydrogen atom and side

**Chain** ®

# **Carbon Numbering System**

Lysine

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Name	Abbreviation	Structural formula*	
Amino acids with nonpolar side chains			
Glycine	Gly (G)	<sup>+</sup> NH₃ 	
Alanine	Ala (A)	NH₃   H₃C—CHCO₂ <sup>−</sup>	
Valine <sup>†</sup>	Val (V)	$\stackrel{NH_3}{\mid}$ $(CH_3)_2CH - CHCO_2^-$	
Leucine <sup>†</sup>	Leu (L)	$(CH_3)_2CHCH_2$ — $CHCO_2$	
Isoleucine <sup>†</sup>	lle (I)	CH <sub>3</sub> NH <sub>3</sub>       CH <sub>3</sub> CH <sub>2</sub> CH—CHCO <sub>2</sub>	
Methionine <sup>†</sup>	Met (M)	CH <sub>3</sub> SCH <sub>2</sub> CH <sub>2</sub> —CHCO <sub>2</sub> —	
Proline	Pro (P)	$H_2C$ $\uparrow$ $H_2$ $H_2C$ $\downarrow$	
Phenylalanine <sup>†</sup>	Phe (F)	~	
Tryptophan <sup>†</sup>	Trp (W)	CH <sub>2</sub> —CHCO <sub>2</sub> -	
Amino acids with polar but nonionized side chains			

Asn (N) Asparagine H<sub>2</sub>NCCH<sub>2</sub>—CHCO<sub>2</sub>

(Continued)

Table 27.1  $\alpha$ -Amino Acids Found in Proteins (Continued)

Name	Abbreviation	Structural formula*		
Amino acids with polar but nonionized side chains				
		O NH <sub>3</sub>		
Glutamine	Gln (Q)	O NH <sub>3</sub>      H <sub>2</sub> NCCH <sub>2</sub> CH <sub>2</sub> —CHCO <sub>2</sub> -		
		NH <sub>3</sub>		
Serine	Ser (S)	HOCH <sub>2</sub> —CHCO <sub>2</sub> -		
		OH ŅH₃		
Threonine <sup>†</sup>	Thr (T)	CH₃CH—CHCO₂¯		
		, NH <sub>3</sub>		
Tyrosine	Tyr (Y)	HO—CH <sub>2</sub> —CHCO <sub>2</sub>		
		<sup>+</sup> NH₃		
Cysteine	Cys (C)	HSCH <sub>2</sub> —CHCO <sub>2</sub> -		
Amino acids with acidic	side chains			
		O		
Aspartic acid	Asp (D)	OCCH <sub>2</sub> —CHCO <sub>2</sub>		
		O NH <sub>3</sub>		
Glutamic acid	Glu (E)	OCCH <sub>2</sub> CH <sub>2</sub> —CHCO <sub>2</sub>		
Amino acids with basic s	ide chains			
		NH <sub>3</sub>		
Lysine <sup>†</sup>	Lys (K)	H <sub>3</sub> NCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> —CHCO <sub>2</sub> —		
		$\stackrel{+}{\text{NH}}_2$ $\stackrel{+}{\text{NH}}_3$		
Arginine <sup>†</sup>	Arg (R)	H <sub>2</sub> NCNHCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> —CHCO <sub>2</sub>		
		, NH³		
Histidine <sup>†</sup>	His (H)	N CH <sub>2</sub> -CHCO <sub>2</sub> -		
		N H		

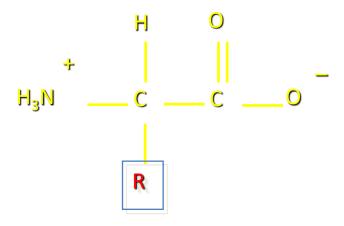
<sup>\*</sup>All amino acids are shown in the form present in greatest concentration at pH 7.  $^\dagger$ An essential amino acid, which must be present in the diet of animals to ensure normal growth.

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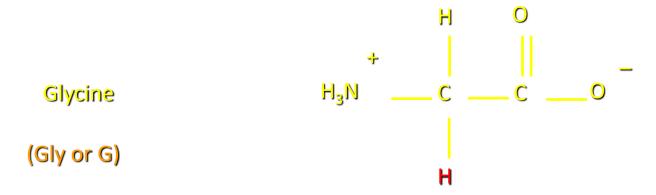
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#### $\alpha$ -Amino Acids Found in Proteins *(Continued)*

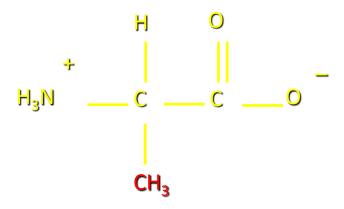
Name	Abbreviation	Structural formula*		
Amino acids with polar but nonionized side chains				
		Ö ŅH³		
Glutamine	Gln (Q)	$\begin{array}{ccc} O & & \stackrel{\leftarrow}{NH_3} \\ \parallel & & \downarrow \\ H_2NCCH_2CH_2 & & CHCO_2^- \end{array}$		
		, NH3		
Serine	Ser (S)	HOCH <sub>2</sub> —CHCO <sub>2</sub> -		
	5-1 (0)			
_, ,		OH NH <sub>3</sub>     CH <sub>3</sub> CH—CHCO <sub>2</sub>		
Threonine <sup>†</sup>	Thr (T)	CH₃CH—CHCO₂		
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Tyrosine	Tyr (Y)	$HO \longrightarrow CH_2 - CHCO_2^-$		
		+		
		NH <sub>3</sub>		
Cysteine	Cys (C)	HSCH <sub>2</sub> —CHCO <sub>2</sub>		
Amino acids with acidic side chains				
		O NH <sub>3</sub>       COCCH <sub>2</sub> —CHCO <sub>2</sub> —		
Aspartic acid	Asp (D)	¯OCCH <sub>2</sub> —CHCO <sub>2</sub> ¯		
		O NH <sub>3</sub>		
Glutamic acid	Glu (E)	O NH <sub>3</sub>         -OCCH <sub>2</sub> CH <sub>2</sub> —CHCO <sub>2</sub> -		
Amino acids with basic side chains				
		NH <sub>3</sub>		
Lysine <sup>†</sup>	Lys (K)	H <sub>3</sub> NCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> —CHCO <sub>2</sub>		
		, NH <sub>2</sub> , NH <sub>3</sub>		
Arginine <sup>†</sup>	Arg (R)	H <sub>2</sub> NCNHCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> —CHCO <sub>2</sub> —		
	J . ,	4		
re-e-et	115-710	NH <sub>3</sub> 		
Histidine <sup>†</sup>	His (H)	CH <sub>2</sub> —CHCO <sub>2</sub>		
		Н		



- The amino acids obtained by hydrolysis of proteins differ in respect to R (the side chain).
- The properties of the amino acid vary as the structure of **R** varies.

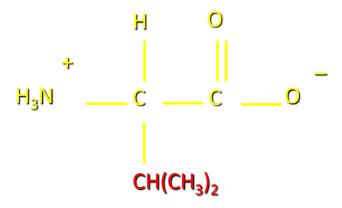


- Glycine is the simplest amino acid. It is the only one in the table that is achiral.
- In all of the other amino acids in the table the  $\alpha$  carbon is a stereogenic center.



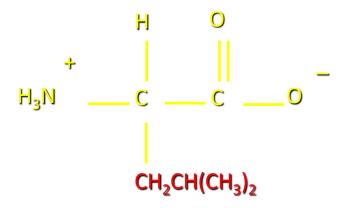
**Alanine** 

(Ala or A)



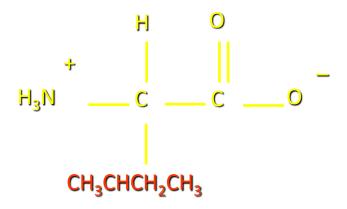
Valine

(Val or V)



Leucine

(Leu or L)

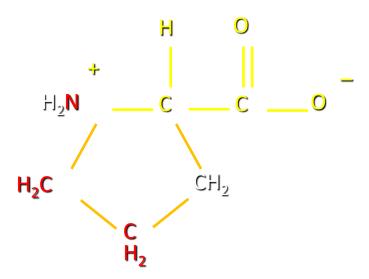


Isoleucine

(Ile or I)

Methionine

(Met or M)

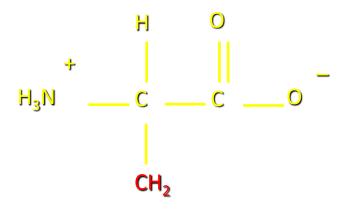


**Proline** 

(Pro or P)

Phenylalanine

(Phe or F)



N (Trp or W)

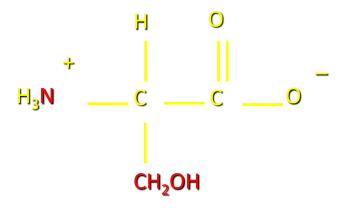
H

Asparagine

(Asn or N)

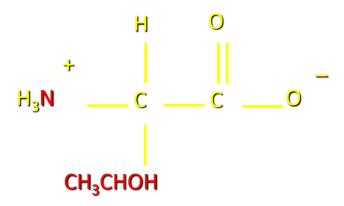
Glutamine

(Gln or Q)



Serine

(Ser or S)



Threonine

(Thr or T)

Aspartic Acid

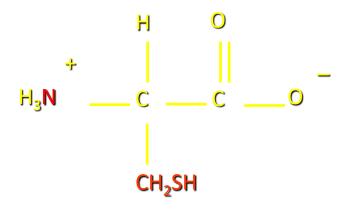
(Asp or D)

**Glutamic Acid** 

(Glu or E)

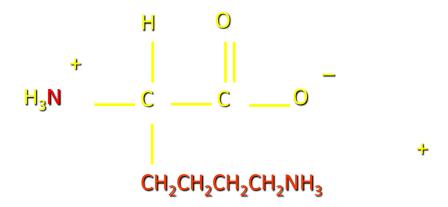
Tyrosine

(Tyr or Y)



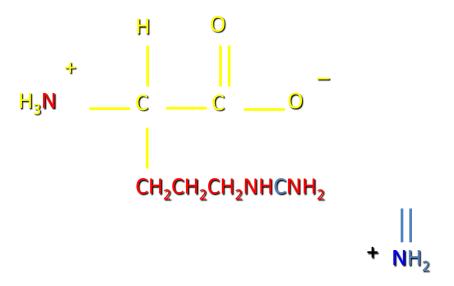
Cysteine

(Cys or C)



Lysine

(Lys or K)



**Arginine** 

(Arg or R)

N NH

Histidine

(His or H)

# Configuration of $\alpha$ -Amino Acids

• Glycine is achiral. All of the other amino acids in proteins have the L-configuration at their  $\alpha$  carbon.

